## Geochemical Monitoring of Fluid-rock Interaction and CO<sub>2</sub> Storage at the Weyburn CO<sub>2</sub>-injection Enhanced Oil Recovery Site, Saskatchewan, Canada

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The Weyburn Field, Saskatchewan, in the Mississippian Midale Formation, is the site of a large CO<sub>2</sub> injection project. Primary and secondary recovery is thought to have depleted the Midale vuggy zone, leaving the marly with higher oil saturations. PanCanadian Resources began tertiary recovery by injection of CO<sub>2</sub> in October 2000. This presented an opportunity to study the potential for geological storage of CO<sub>2</sub> as established by the multidisciplinary IEA Weyburn CO<sub>2</sub> program of which geochemical modeling plays an important role.

Comparing pre (baseline) and post injection (monitor) samples has recognized changes in the fluid chemistry and isotope composition. The distribution of CI, Ca, NH<sub>3</sub>,  $\delta^{13}$ CH<sub>4</sub> and  $\delta^{13}$ CO<sub>2</sub> increase from northwest to southeast, and alkalinity, resistivity, H<sub>2</sub>S and  $\delta^{34}$ SO<sub>4</sub> decrease. After injection of CO<sub>2</sub>, the general patterns are similar, but pH has dropped by 0.5 units and alkalinity has doubled, consistent with calcite dissolution due to reaction with injected CO<sub>2</sub>. Isotope data support this interpretation. Baseline samples varied from -22 to -12‰  $\delta^{13}$ C (V-PDB) for CO<sub>2</sub> gas. The injected CO<sub>2</sub> has an isotope ratio of -34.5 ‰. The monitor samples range from -18 to -13‰, requiring a heavy source of carbon for CO<sub>2</sub>, most easily attributed to dissolution of carbonate minerals.

Addition of  $CO_2$  causes dissolution of carbonates and production of  $CO_2$ . Geological storage of  $CO_2$  relies on silicate minerals to buffer pH, causing  $CO_2$  to be precipitated as calcite. Modeling of water-rock reactions suggests that clay minerals may be present and capable of acting as pH buffers, allowing injected  $CO_2$  to be stored for geological time as carbonate minerals.